

Solar photocatalytic H₂ production over CeO₂-based catalysts

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Introduction and Methods

Photocatalytic approach is a green and alternative process to produce hydrogen via water splitting¹. In photocatalysis, CeO₂ is a less active semiconductor than TiO₂ (the most used photocatalyst), but its lower band gap allows it to adsorb a major portion of visible light¹. One of the most employed strategy for this purpose, is the synthesis of nanocomposites with other oxides. WO₃ is widely used as holes donor in water splitting processes and its combination with CeO₂ could be a suitable strategy to increase the H₂ yield². Recently, the photo-reforming process of an organic compound is also considered as a good approach³, and that is why in the reaction mixture an alcohol is often added as sacrificial agent³. In this context, the use of a bio-alcohol could be an additional advantage. In this work, were synthesized several composites using two different preparation methods: co-precipitation and templated synthesis. In particular, were employed two different templating agents, the cetrimonium bromide (CTAB) and the hexamethylenetetramine (HMTA). For each preparation, a 5% wt and 10% wt of tungsten oxide were added, for a total of six different samples. Finally, the most performant sample were treated with a pulsing laser, in order to modify its surface properties.

Results

In the results, it is possible to note how a low percentage of WO₃ enhances the activity in co-precipitated samples (CeO₂-W5) respect to bare CeO₂. However, this behaviour is opposite in HMTA templated synthesis, where the best activity among all untreated composites is given by the 10% modified sample (CeO₂-W10 [HMTA]). Instead, CTAB preparation shows the lowest activity. Looking the treated samples, it is interesting as the laser treatment enhances the activity more in the composite than in the bare sample.

Conclusion

CeO₂-WO₃ composites were prepared as unconventional photocatalysts for hydrogen evolution.

The 10% wt HMTA templated synthesis gave the best results with 0.175 mmol/g h of H₂. Furthermore, the laser treatment induces defective centres on the material surface that increase the H₂ yield more than two times (0.4 mmol/g h), how verified also by Raman spectroscopy and TEM images.

References

1. Fiorenza, R. *et al.* Au/TiO₂-CeO₂ catalysts for photocatalytic water splitting and VOCs oxidation reactions. *Catalysts* **6**, 121 (2016).
2. Su, J., Guo, L., Bao, N. & Grimes, C. A. Nanostructured WO₃/BiVO₄ Heterojunction Films for Efficient Photoelectrochemical Water Splitting. *Nano Lett.* **11**, 1928–1933 (2011).
3. Bowker, M. *et al.* The photocatalytic window: Photo-reforming of organics and water splitting for sustainable hydrogen production. *Catal. Letters* **145**, 214–219 (2015).

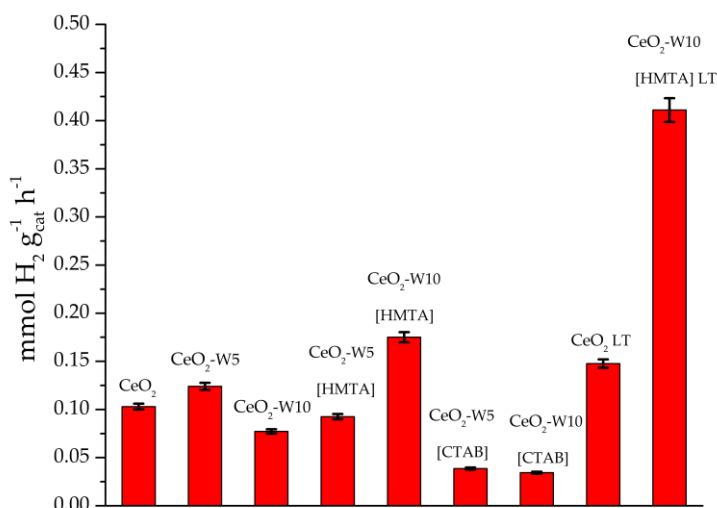


Figure 1 H₂ production for each sample