

Operando FT-IR analysis for methane coupling reaction to produce C2 chemicals

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Highlights

- Non-oxidative methane coupling was conducted in a DBD plasma reaction cell
- Methane coupling reaction was analyzed via operando FT-IR method
- Characteristic peaks of intermediate species were identified and analyzed

1. Introduction

As shown in Figure 1, we used a dielectric barrier discharge (DBD) plasma reactor cell made with insulating material and ZnSe window for an operando Fourier transform infrared (FT-IR) spectroscopy. The analysis was conducted at a reduced pressure and near room temperature. The ordered mesoporous materials (OMMs) or OMM-supported catalysts mixed with inorganic sol was coated on the ZnSe window for the operando analysis (Figure 2). During the DBD plasma reaction, absorption spectra of FT-IR were recorded about types of activated and adsorbed hydrocarbon intermediates. Through this analysis, key species for coupling reaction were identified, and the difference was also observed with or without catalytic material. In addition, the influence of plasma on the surface characteristics of particles as well as catalytic nanocrystals was investigated. Through this operando technology, the performances of plasma and catalyst were successfully assessed.

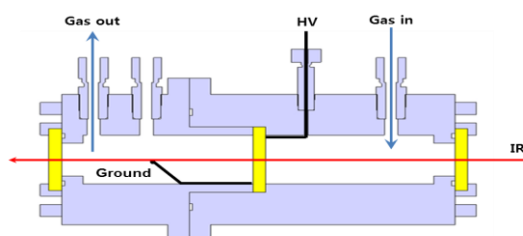


Figure 1. Reaction cell with plasma for FT-IR analysis

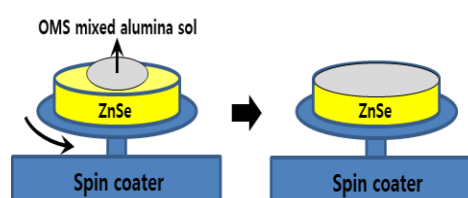


Figure 2. Spin-coated catalysts

2. Methods

OMS particles were prepared by the method explained elsewhere [1]. The structure directing agent P123 ($\text{EO}_{20}\text{PO}_{70}\text{EO}_{20}$, MW = 5800 g/mol) was mixed with aqueous HCl solution. Next, ethanol was added to the mixture and then TEOS (tetraethyl orthosilicate) was slowly added followed by mixing for 24 h. At 100 °C, the mixture was hydrothermally synthesized, followed by washing, drying and calcination. During the analysis, a feed mixture (methane 50%) was introduced to the cell under plasma. The physical properties were shown in Table 1 and Figure 3.

Sample	BET Surface Area (m^2/g)	Pore Volume (cm^3/g)	Pore Diameter (nm)
OMS	697	1.00	6.27

Table 1. Physisorption result of OMS

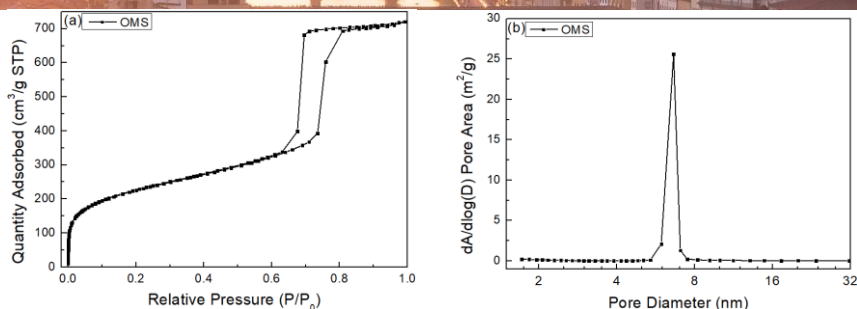


Figure 3. Physisorption result of OMS; (a) N₂ adsorption-desorption isotherm (b) pore size distribution

3. Results and discussion

In Figure 4, the peaks involved with CH₄ molecules only were observed before reaction. [2] Due to the reaction under plasma, new peaks closely related to the formation of C₂ chemicals (ethane, ethylene and acetylene) appeared. [3] At 2800 – 3000 cm⁻¹, stretching vibration of CH₃ species was observed, and these species are highly related to the formation of ethane. The peak near 950 cm⁻¹ was reportedly due to out-of-plane vibration, and these species are closely related to the formation of unsaturated C₂ chemical, ethylene. Another peak near 730 cm⁻¹ could be ascribed to the formation of acetylene, also one of the unsaturated C₂ hydrocarbons. By analyzing intermediate hydrocarbon species as shown in Figure 4, the selectivities of C₂H₆, C₂H₄, C₂H₂ as well as reaction performance could be successfully predicted.

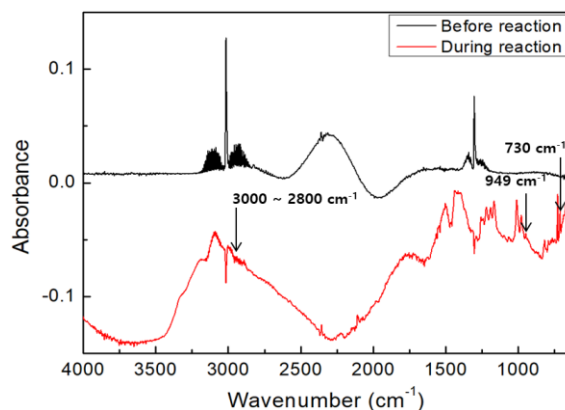


Figure 4. FT-IR spectra during reaction

4. Conclusions

By employing the operando FT-IR spectroscopy, key intermediate components were easily identified and reaction pathway of each C₂ component could be predicted. Furthermore, catalytic or non-catalytic reaction in the DBD plasma cell will be tested in order to find the role of catalyst in the DBD plasma bed.

References

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Keywords

“Operando FT-IR analysis”, “Ordered mesoporous silica”, “Dielectric barrier discharge plasma”, “Methane coupling”