

Modelling of falling film reactors for exothermic reactions

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Anionic surfactants are the key component in a detergent formulation. These molecules are composed of a lipophilic “tail” (typically C12-C14) and a hydrophilic “head” (i.e. SO_3^-). Organic molecules obtained from renewable sources, such as vegetable oils, are currently used as raw materials. The synthesis is normally performed in continuous falling-film reactors where SO_3 is dissolved in a thin film of organic compound. The reaction is characterized by high exothermic tenor ($\Delta H=-150\text{kJ/mol}$), thus a critical aspect deals with the heat exchange system. For this reason, the reaction is typically performed in multi-tubular systems, characterized by very efficient heat exchange. Modelling such a reactor is a hard topic to be faced, because a rigorous model should consider both the heat and mass-transfer phenomena in both gas and liquid phases. By considering that normally the liquid film flows under laminar regime, fact due to the high residence time necessary to achieve high conversion, the system increases in complexity, because it is necessary to implement the mass diffusion and heat transfer contributes also in the radial coordinate of the reactor. In the present work, a detailed mathematical model is proposed and dodecyl benzene sulfonation has been chosen as a case study [1,2]. The breakthrough idea is to develop a model where no rate-determining-step is considered, writing the opportune mass and heat balance equations on both the gas and liquid phases.

A parametric study was performed by fixing the experimental conditions adopted in the literature ($L=2\text{m}$, $F_{\text{SO}_3,\text{feed}}=2.2 \cdot 10^{-3}\text{mol/s}$, $\text{SO}_{3,\text{feed}}=4\%$, $\text{SO}_3/\text{A}=1.1\text{mol/mol}$, $T_{\text{feed}}=320\text{K}$) [3] and the kinetic rate laws with related parameters [3]. As an example, the result of the substrate conversion is reported in Figure 1A, while the liquid temperature in Figure 1B.

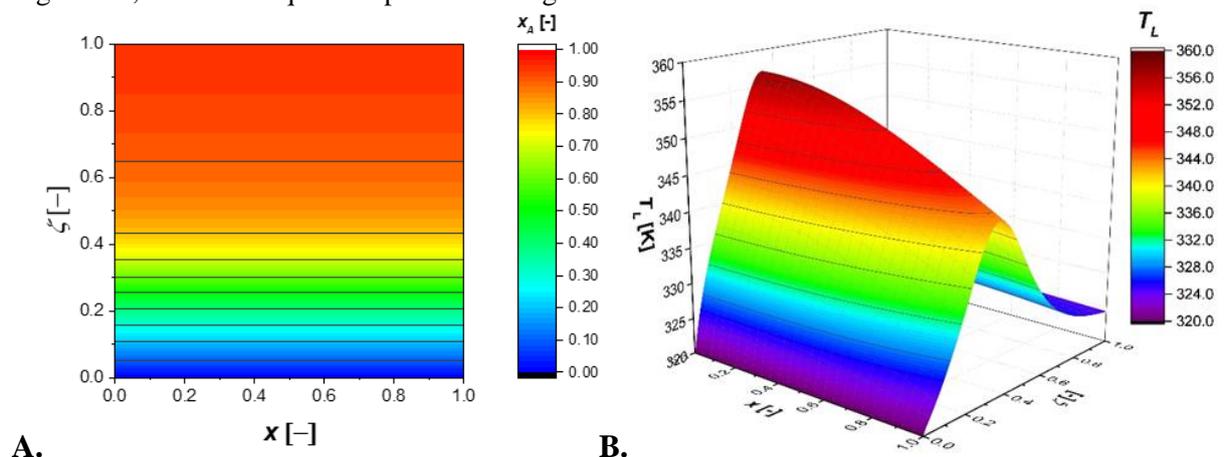


Figure 1. A. Conversion of the organic substrate as a function of the axial dimensionless coordinate (ξ) and parametric with the radial dimensionless coordinate (y). B. Dimensionless liquid temperature variation with the axial dimensionless coordinate (ξ) and the radial dimensionless coordinate (y).

The trends are reliable, and the conversion shows a profile in the liquid film radial coordinate. The film thickness decreases with the conversion degree, fact that strictly depends on the physical properties of the substance to be sulfonated.

References

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