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# Sustainable Combined Production of Hydrogen and Energy from Biomass in Malaysia

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This work relates to a comparison between process solutions for the production of  $H_2$  and the coproduction of  $H_2$  and energy by means of a zero emission biomass integrated gasification and a combined cycle (BIGCC) power plant. The energy production is 10 MWe, in agreement with the Small Renewable Energy Power Plant (SREP) Program, promoted by the Government of Malaysia.  $H_2$  is obtained by supercritical water gasification (SCWG), a technology of interest for the processing of biomass with high moisture content. An economic analysis has been carried out in order to demonstrate the feasibility of the process solutions and to compare their convenience. The feedstock is 280,000 t/y of Empty Fruit Bunch (EFB), a biomass obtained in the Palm Oil Industry. The location of the site is Teluk Intak District in the State of Perak (Malaysia). The processes are designed with Aspen Plus<sup>®</sup> V7.2. The aim of this work is to develop detailed process flow diagrams for the supercritical water gasification technology in order to study and compare the convenience and the sustainability of different scenarios that can be adopted in an industrial context. The processes have been developed to reach zero emissions and zero wastes. CO<sub>2</sub> and solid residuals are recycled inside the palm oil lifecycle. A cost analysis has been performed to find out the convenience of the proposed solutions.

## 1. Introduction

Environmental issues and large usage of fossil fuels have led the Government of Malaysia to develop an energy plan which is aiming to reduce 40 % of its greenhouse gases emissions by 2020. To achieve this goal Malaysia has promoted the usage of renewable sources in power generation, launching the Small Renewable Energy Power Plant Program in May 2001. Power producers can use every kind of renewable sources in energy production with an income tax exemption of 70 % on the statutory income for five years (EC-ASEAN, 2002).

Since Malaysia has a significant amount of agricultural activities and is one of the leading manufacturers and exporters of palm oil worldwide, biomass is a very promising alternative source of renewable energy in Malaysia's energy policy (Mohammed et al., 2011; Sumathi et al., 2008).

In 2008 the Malaysian land covered with palm oil plantations was 4.5 Mha (MPOB, APOC, 2010). It has been estimated by Kelly-Yong et al. (2007) that a hectare of palm oil can yield 21.625 t/y of biomass residues and Empty Fruit Bunches (EFB) are the 20.44 %. Biomass from palm oil industries also shows great reactivity due to its chemical composition. It can be used as a raw material for the production of a large variety of chemicals: biofuels, biodiesel, resins, bioethanol, fertilizers and H<sub>2</sub>, although also ashes and wastes obtained during palm oil biomass processing can be reused as soil conditioners or fertilizers, due to their good content of micro and macro nutrients (Shuit et al., 2009).

 $H_2$  is widely used in different industrial sectors already present in Malaysia: petroleum refining, petroleum chemistry and food processing.

The aim of this work is to develop detailed process flow diagrams for the SCWG technology in order to study and compare the convenience and the sustainability of different solutions that can be adopted in an industrial scenario for the production of  $H_2$  and the co-production of  $H_2$  and 10 MWe of energy, according

to the Malaysia SREP project, on the basis of a real palm oil plantation (Maju Intan Biomass Energy Power Plant Project, 2009).

## 2. SCWG operating conditions

Before simulating the scenarios, a preliminary study on the effect of operating conditions (temperature, pressure and water to biomass ratio) for the SCWG reaction has been performed.

The problem has been carried out using the Gibbs free energy minimization method (Tang and Kitagawa, 2005; Castello and Fiori, 2011; Freitas and Guirardello, 2012) and the Peng-Robinson equation of state, since it shows a good agreement with the experimental data (Castello and Fiori, 2011).

The characterization of the EFB biomass has been taken from the work by Omar et al. (2011). The moisture content has been set to 60 % by weight (Maju Intan Biomass Energy Power Plant Project, 2009). The results are reported in Figure 1.

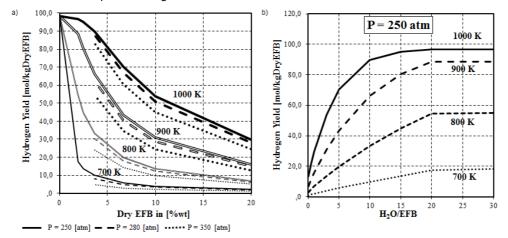


Figure 1. a) Effect of pressure, temperature and EFB biomass dry content in the reactor's feed on the  $H_2$  yield. b) Effect of temperature and water/EFB ratio in the reactor's feed on the  $H_2$  yield at 250 [atm].

The obtained results show that  $H_2$  yield decreases with pressure and increases with temperature. At a given pressure and temperature, the  $H_2$  yield of reaction decreases with the biomass dry content of the reactor's feed. The highest yields are realized at 1,000 K and a subsequent temperature increase does not affect  $H_2$  yield anymore (Lu et al., 2007). 250 atm is the pressure level that guarantees the highest  $H_2$  yield. A water/EFB ratio of 10 (3.6 – 4 % by weight of dry EFB content) has been selected for the SCWG reactor, since a higher value for this parameter does not lead to a significant change in the  $H_2$  yield of reaction.

# 3. Process description and simulation

Two possible cases are studied: the  $H_2$  production without energy sales and the  $H_2$  production coupled with 10 MWe of energy sold to the Malaysia National Grid.

The proposed scenarios are four and they can be described using two process flow diagrams (PFDs) (Figure 2). In the first PFD a combined-cycle power plant coupled with the H<sub>2</sub> production plant is proposed, whereas in the second PFD a turbogas plant is used for the energy production.

35,000 kg/h of EFB (60 % by weight moisture content) are fed to a crusher and mixed with water to obtain a slurry. The slurry is then pumped above the water's critical pressure, heated and fed to the gasifier. Ashes obtained from the process are separated from the product gas. The obtained raw syngas is cooled to be purified. Heat recovery between the produced gas from the reactor and the inlet feed is required to improve the thermodynamic efficiency of the process. The efficiency of the heat recovery is 96.61 %. Since the SCWG reaction is endothermic, 4,076.71 kg/h of additional EFB are necessary to completely satisfy the reaction's heat requirements. The gas is then cooled to 38 °C and expanded to 37 bar (UOP, 2009) for the removal of acidic components (CO<sub>2</sub> and H<sub>2</sub>S). Water and condensates are subsequently expanded to the atmosphere, mixed with other sour water streams and fed to the sour water stripper to recovery and recycle the water into the process. The syngas from the first flash drum is sent to the Selexol<sup>TM</sup> unit, while the acid gas from the atmospheric flash drum is mixed with other sulfur-containing gas streams and sent to

608

the Claus unit for sulfur recovery. In the first column of the Selexol<sup>TM</sup> plant H<sub>2</sub>S removal occurs, while in a second absorption column CO<sub>2</sub> is captured. The Selexol<sup>TM</sup> solvent is then regenerated. For the H<sub>2</sub>S section a distillation column is used, while for the CO<sub>2</sub> section an air stripper is adopted (Sweney, 1973; Burr and Lyddon, 2008). The obtained H<sub>2</sub>S is mixed with the streams that come from both the atmospheric flash drum and the sour water stripper to be fed to the Claus unit (Kohl and Nielsen, 1997).

The H<sub>2</sub> from the Selexol<sup>TM</sup> unit is divided into two streams: the former is used as fuel, whereas the latter is upgraded in a PSA unit. Pure H<sub>2</sub> is ready for the market, while impurities are compressed again to 37 bar and mixed with the H<sub>2</sub> fuel stream.

The two configurations adopted differ in the energy production plant. In the first scheme (Figure 2b) a combined-cycle power plant is used:  $H_2$  is diluted with  $CO_2$  (Natarajan et al., 2007), burned and expanded in a gas turbine. Flue gases are sent to the DeNOx-SCR reactor (NOx and SOx removal) and the heat of the clean flue gases is used for steam production to drive a steam cycle with three pressure levels (150, 30, 5 bar) and as utility (1.5 bar steam) for the plant.

In the second scheme (Figure 2a) only a gas turbine is used for power generation and the heat of the flue gases is used only to produce low pressure steam (1.5 bar) as utility.

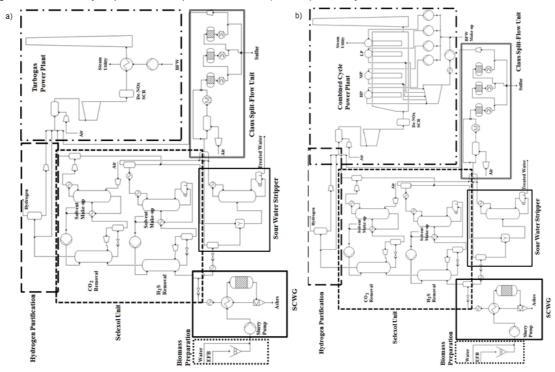


Figure 2. PFDs for the production of  $H_2$  and the co-production of  $H_2$  and energy, using a) a turbogas power plant and b) a combined-cycle power plant.

When only  $H_2$  is produced, the fraction of  $H_2$  burned is reduced to satisfy only the plant requirements (heat and power), while when the production of  $H_2$  is coupled with the production of 10 MWe of energy, this ratio is increased to both satisfy the plant's energy requirements and produce the desired amount of electrical power. For the co-production of  $H_2$  and 10 MWe of power with a combined-cycle power plant (scenario 1) the  $H_2$  stream used as fuel is 41.8% of the total gas stream coming out from the Selexol<sup>TM</sup> unit. For the production of only  $H_2$  with the fulfillment of the plant's energy requirements with a combined-cycle power plant (scenario 2), the value of this parameter is 24.3%. For the co-production of  $H_2$  and 10 MWe of power using a gas turbine (scenario 3), the  $H_2$  used as fuel is the 51% of the total stream from Selexol<sup>TM</sup> unit. At last, for the production of only  $H_2$  and energy to satisfy the plant's requirements (scenario 4), the value of this parameter is 18.1%.

Pumps isentropic efficiencies have been set to 75 %, compressors isentropic efficiencies are equal to 80 %. Gas turbine isentropic efficiency is 90 %, whereas for the steam cycle the efficiencies of the high pressure stage, medium pressure stage and low pressure stage are respectively 84 %, 89 % and 77 % (Aspen Plus<sup>®</sup>, 2008). To better perform the simulations, also machineries drivers efficiencies and parasitic

load consumptions have been taken into consideration. Compressor drivers, pump drivers and EFB mill have an efficiency of 92 % (Schiferl, 2003), while the alternator 95 % (Pellegrini et al., 2011). For parasitic loads the following assumptions have been considered: steam turbine auxiliary equipment consume the 0.1 % of the gross electric power produced by the steam turbine (Gülen, 2011), service water pump and De-NOx SCR circulating pump require the 0.1 % of the gross electrical power of the combined cycle (ABB, 2009), the plant instrumentation and transformer need the 0.5 % of the gross electrical power of the combined cycle (ABB, 2009) and the gas turbine auxiliary equipment absorbs the 0.25 % of the gross electrical power from gas turbine (Gülen, 2011). The belt conveyor for biomass transport requires 1.5 kWe (Clénet, 2010). For scenario 1 and scenario 2 a 0.1% loss of the total BFW used in the steam cycle has been considered (Pellegrini et al., 2011). The plant operability is 8,000 h/y.

### 4. Economic and Sustainability Analysis Assumptions

To make a comparison between the four scenarios considered in this work, a cost analysis has been performed, taking into account both capital and operating costs. Capital costs have been calculated according to Eq. (1):

$$C = \frac{M\&S2011}{M\&Syear} C_0 \left(\frac{S}{S_0}\right)^n \tag{1}$$

where *C* is the actual cost of the plant's units,  $C_0$  is the reference cost. *S* is the actual size of every process unit and  $S_0$  is the reference size. *n* is the scale factor and *M&S2011*, *M&Syear* are the Marshall and Swift indexes respectively at year 2011 and at the year at which the reference unit cost is given. The reference costs and sizes and the scale factor are taken from the work by Kreutz et al. (2005) and General Atomics (2005). General facilities, contingencies, land, site specific factor, engineering, permits and start-up costs have been calculated as percentages of total capital costs (Pellegrini et al., 2011).

Operating costs have been calculated assuming 0.001 \$/kg for BFW (Pellegrini et al., 2011), 4.32 \$/kg for the Selexol<sup>TM</sup> solvent (Rubin et al., 2007), whereas EFB cost has been considered zero, since the plant is thought to be installed next to the palm oil plantations and mills.

For revenues, electricity selling price has been considered 0.062 kWh (UNDP, 2007), H<sub>2</sub> value is 0.84 kg (Kelly-Yong et al., 2007) and the price of sulfur is 0.285 kg (Alibaba, 2013).

The economical comparison between the four studied scenarios has been carried out evaluating the payback time (assumed as the ratio between capital costs and the difference between revenues and operating costs).

The environmental sustainability of the proposed plant solutions has been investigated. Palm oil is established to be a very sustainable plantation and it is considered capable to have a high efficiency in  $CO_2$  emissions mitigation (Sumathi et al., 2008; Shuit et al., 2009; UNDP, 2007). Palm Oil cultivations can absorb 44 t of dry matter per hectare per year and up to 64.5 t of  $CO_2$  per hectare per year, while typical rainforests can only absorb 25.7 t of dry matter per hectare per year and 42.2 t of  $CO_2$  per hectare per year (Sumathi et al., 2008; Shiut et al., 2009). It has been estimated by Kelly-Yong et al. (2007) that a hectare of palm oil can yield 21.625 t/y of biomass residues and Empty Fruit Bunches (EFB) are 20.44 % of the total biomass produced from palm oil wastes. The environmental sustainability of the four case studies has been discussed in terms of land requirements for the  $CO_2$  and ashes reabsorption by the palm oil cultivations in the area where the four scenarios are located.

## 5. Results and Discussion

The results of the four process simulations are summarized in Table 1. The plant overall efficiency is calculated according to Eq. (2):

$$\eta = \frac{W_{el,NET} + W_{H_2,PROD}}{\dot{W}_{IN,EFB}}$$
(2)

as the ratio between the total power output (electrical and contained in the produced  $H_2$ ) and the EFB biomass input power.

610

For the evaluation of the revenues, the excess heat produced has not been taken into account, since the aim of these scenarios is to produce only  $H_2$  and/or, electrical power.

	SCENARIO1	SCENARIO2	SCENARIO3	SCENARIO4
H <sub>2</sub> Production [t/y]	11671.29	15180.71	9826.35	16424.04
Electric Power Requirement [MWe]	16.86	17.40	22.76	14.38
Gross Power Production [MWe]	26.84	17.40	32.83	14.38
Net Power Production [MWe]	10.08	0.0	10.07	0.0
Thermal Requirement [MW]	44.97	45.0	45.0	45.0
Thermal Production [MW]	44.97	45.0	120.0	45.0
Net Thermal Production [MW]	0.0	0.0	75.0	0.0
Plant Overall Efficiency [%]	41.4	44.7	36.0	48.25
Investment Costs [M\$]	65.8665	60.0199	58.1042	47.4123
Income [M\$/y]	13.484	11.451	11.915	12.495
Payout Time [y]	4.88	5.24	4.88	3.79
CO <sub>2</sub> Produced [t/y]	189,313.74	190,543.28	190,318.54	189,313.74
Ashes Produced [t/y]	9,428.49	9,428.49	9,428.49	9,428.49
Land Covered by The Palm Oil Plantation [ha]	70,724.67	70,724.67	70,724.67	70,724.67
Land Required for CO <sub>2</sub> Absorption [ha]	2,935.10	2,954.16	2,950.67	2,935.10
Land Required for Ashes Absorption [ha]	214.28	214.28	214.28	214.28

Table 1: Results obtained in this work

The plant overall efficiency is usually higher when only  $H_2$  is produced and scenario 4 allows to reach the highest value. In scenario 3 a surplus of heat is produced, since the heat of the flue gas from the turbine is not completely recovered and losses are significant, while in scenario 4 the amount of fuel burned for energy production is lower and the excess heat of the flue gas is recovered completely as steam, balancing the plant's thermal requirements. Scenario 1 has the highest investment costs. In scenario 4 the cost for the CO<sub>2</sub> compressor is different, since the amount of CO<sub>2</sub> used for dilution inside the turbine is lower due to the lowest consumption of  $H_2$  as fuel.

The comparison shows that for the production of  $H_2$  scenario 4 is the most suitable solution. For the coproduction of  $H_2$  and 10 MWe of energy scenario 1 and scenario 3 have the same payout time. However scenario 1 is the best solution, since in scenario 3 the energy efficiency is lower, a huge amount of heat is not recovered (75 MW) and the produced  $H_2$  is lower.

Results obtained from the sustainability analysis show that the land required for the absorption of ashes and  $CO_2$  is quite the same for the four scenarios and is less than the area covered with oil palms. In this way all the wastes produced during the biomass processing are completely recycled into the EFB life cycle and the four cases are sustainable.

# 6. Conclusions

A comparison between process solutions for the production of  $H_2$  and the co-production of  $H_2$  and 10 MWe energy (in agreement with the Malaysia Small Renewable Energy Power Plant Program) has been made. An economic analysis has been carried out in order to demonstrate the feasibility of the process solutions and to compare their convenience. The obtained results show that, when only  $H_2$  is produced, the most suitable solution is the coupling of the SCWG plant with a gas turbine for the production of plant utilities, whereas for the combined production of  $H_2$  and power an integration of the SCWG plant with a combined-cycle power plant is more convenient. Detailed process flow diagrams for the SCWG technology have been developed and the process simulations have been performed with Aspen Plus<sup>®</sup>. The designed processes can be considered at zero emissions and zero wastes.  $CO_2$  and solid residuals are recycled inside palm oil lifecycle. The study demonstrates the great potential of palm oil biomass as a renewable energy source.

## References

ABB (Asea Brown Boveri), 2009, Power Generation Energy Efficient Design of Auxiliary Systems in Fossil-Fuel Power Plant, ABB Energy Efficiency Handbook, <www.seeei.org.il>, accessed 4.11.2012.

Aspen Plus® v7.2, 2008, Aspen Plus IGCC Model, Aspen Technology, Inc., Burlington, Massachusetts, USA.

- Burr B., Lyddon L., 2008, A Comparison of Physical Solvents for Acid Gas Removal, Bryan Research & Engineering, Bryan, Texas, USA.
- Castello D., Fiori L., 2011, Supercritical Water Gasification of Biomass: Thermodynamic Constraints, Biores. Technol. 102, 7574-7582.
- Clénet D., 2010, Optimising Energy Efficiency of Conveyors, <www.schneider-electric.com>, accessed 15.11.2012.
- EC-ASEAN (European Commission, Association of South East Asian Nations), 2002, Small Renewable Energy Power (SREP) Programme for the Promotion of Renewable Energy Power Generation, <www.ec-asean-greenippnetwork.net>, accessed 20.10.2012.
- Freitas A.C.D., Guirardello R., 2012, Supercritical Water Gasification of Glucose and Cellulose for Hydrogen and Syngas Production, Chemical Engineering Transactions 27, 361-366, DOI: 10.3303/CET1227061.
- General Atomics, 2005, Supercritical Water Partial Oxidation. Phase I Pilot-Scale Testing/Feasibility Study Final Report, <www.fischer-tropsch.org>, accessed 25.10.2012.
- Gülen S.C., 2011. Importance of Auxiliary Power Consumption for Combined Cycle Performance, J. Eng. Gas Turb. Power 133, 041801 1-10, DOI:10.1115/1.4002254.
- Kelly-Yong T.L., Lee K.T., Mohamed A.R., Bhatia, S., 2007, Potential of Hydrogen from Oil Palm Biomass as a Source of Renewable Energy Worldwide, Energy Policy 35, 5692-5701.
- Kohl A., Nielsen R., 1997, Gas Purification, Fifth edition, Gulf Publishing Company, Houston, Texas.
- Kreutz T., Williams R., Consonni S., Chiesa P., 2005, Co-Production of Hydrogen, Electricity and CO<sub>2</sub> from Coal with Commercially Ready Technology. Part B: Economic Analysis, Int. J. Hydrogen Energy 30, 769-784.
- Lu Y.J., Guo L., Zhang X., Yan Q., 2007, Thermodynamic Modeling and Analysis of Biomass Gasification for Hydrogen Poduction in Supercritical Water, Chem. Eng. J. 131, 233-244.
- Maju Intan Biomass Energy Power Plant Project, 2009, <www.eco2data.com>, accessed 19.10.2012.
- Mohammed M.A.A., Salmiaton A., Wan Azlina W.A.K.G., Mohammad Amran M.S., Fakhru'l-Razi A., Taufiq-Yap Y.H, 2011, Hydrogen Rich Gas from Oil Palm Biomass as a Potential Source of Renewable Energy in Malaysia, Renew. Sustain. Energy Rev. 15, 1258-1270.
- MPOB, APOC (Malaysian Palm Oil Board, American Palm Oil Council), 2010, Palm Oil Development and Performance in Malaysia, <www.americanpalmoil.com>, accessed 05.12.2011.
- Natarajan J., Lieuwen T., Seitzman J., 2007, Laminar Flame Speeds of H2/CO Mixtures: Effect of CO<sub>2</sub> Dilution, Preheat Temperature, and Pressure, Combust. Flame 151, 104-119.
- Omar R., Idris A., Yunus R., Khalid K., Aida Isma M.I., 2011, Characterization of Empty Fruit Bunch for Microwave-Assisted Pyrolysis, Fuel 90, 1536-1544.
- Pellegrini L.A., Soave G., Gamba S., Langè S., 2011, Economic Analysis of a Combined Energy–Methanol Production Plant, Appl. Energy 88, 4891-4897.
- Alibaba, <price.alibaba.com/price/priceLeafCategory.htm?spuId=100069569&categoryId=100001646>, accessed 09.01.2013.
- Rubin E.S., Berkenpas M.B., Frey H.C., Chen C., McCoy S.T., Zaremsky C.J., 2007, Technical Documentation: Integrated Gasification Combined Cycle Systems (IGCC) with Carbon Capture and Storage (CCS), Department of Engineering and Public Policy, Paper 76, <repository.cmu.edu/epp/76>, accessed 10.12.2012.
- Schiferl R., 2003, An Accurate Method to Determine Electric Motor Efficiency While the Motor is in Operation, Texas Technology Showcase, Energy-Efficiency in the Chemical and Refining Industries, March 17-19 2003, Houston, Texas, USA.
- Shuit S.H., Tan K.T., Lee K.T., Kamaruddin A.H., 2009, Oil Palm Biomass as a Sustainable Energy Source: A Malaysian Case Study, Energy 34, 1225-1235.
- Sumathi S., Chai S.P., Mohamed A.R., 2008, Utilization of Oil Palm as a source of Renewable Energy in Malaysia, Renew. Sustain. Energy Rev. 12, 2404-2421.
- Sweney J.W., 1973, Synthetic Fuel Gas Purification by the Selexol<sup>®</sup> Process. ACS Fuel Volumes, Spring 1973, Dallas, USA, 142-156, <web.anl.gov/PCS/ENFL>, accessed 5.12.2012.
- Tang H., Kitagawa K., 2005, Supercritical Water Gasification of Biomass: Thermodynamic Analysis with Direct Gibbs Free Energy Minimization, Chem. Eng. J. 106, 261-267.
- UNDP Malaysia (United Nations Development Programme), 2007, Malaysia Generates Renewable Energy from Palm Oil Wastes, <www.undp.org.my>, accessed 5.12.2012.
- UOP (Universal Oil Products), 2009, UOP Selexol<sup>™</sup> Technology for Acid Gas Removal, <www.uop.com>, accessed 20.11.2012.